

18.30 (a) This germanium material to which has been added  $10^{24} \text{ m}^{-3}$  As atoms is  $n$ -type since As is a donor in Ge. (Arsenic is from group VA of the periodic table--Ge is from group IVA.)

(b) Since this material is  $n$ -type extrinsic, Equation 18.16 is valid. Furthermore, each As atom will donate a single electron, or the electron concentration is equal to the As concentration since all of the As atoms are ionized at room temperature; that is  $n = 10^{24} \text{ m}^{-3}$ , and, as given in the problem statement,  $\mu_e = 0.1 \text{ m}^2/\text{V}\cdot\text{s}$ . Thus

$$\begin{aligned}\sigma &= n |e| \mu_e \\ &= (10^{24} \text{ m}^{-3})(1.602 \times 10^{-19} \text{ C})(0.1 \text{ m}^2/\text{V}\cdot\text{s}) \\ &= 1.6 \times 10^4 (\Omega\cdot\text{m})^{-1}\end{aligned}$$